**In vitro Effect of Lead, Silver, Tin, Mercury, Indium and Bismuth on Human Sperm Creatine Kinase Activity: a Presumable Mechanism for Men Infertility**

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**ABSTRACT**

**Background:** The aim of the present study was to investigate the *in vitro* effects of mercury (Hg\(^{2+}\)), lead (Pb\(^{2+}\)), silver (Ag\(^{2+}\)), tin (Sn\(^{2+}\)), bismuth (Bi\(^{3+}\)) and indium (In\(^{3+}\)) ions on sperm creatine kinase. **Methods:** creatine kinase was isolated from human sperm homogenates after chromatography on a DEAE cellulose column. **Results:** At 60 µg ml\(^{-1}\) metal concentration, 70% of the creatine kinase activity was inhibited by Hg\(^{2+}\), while at the same concentration, Pb\(^{2+}\), Ag\(^{2+}\), Sn\(^{2+}\), Bi\(^{3+}\) and In\(^{3+}\) caused 68%, 66.5%, 65.7%, 64.7% and 62.7% inhibition, respectively. All six metal ions displayed a competitive type of inhibition mechanism for the isolated creatine kinase as analyzed by Lineweaver-Burk plot. K\(_i\) values of Hg\(^{2+}\), Pb\(^{2+}\), Ag\(^{2+}\), Sn\(^{2+}\), Bi\(^{3+}\) and In\(^{3+}\) were calculated and 8.34 mM, 5 mM, 4.54 mM, 3.45 mM, 3.12 mM and 2.63 mM values were obtained, respectively. **Conclusion:** All the studied metal ions, at levels of 60 µg ml\(^{-1}\), may reduce normal sperm metabolism by inhibition of sperm creatine kinase, which probably is an important cause of infertility in men. However, further investigations, as *in vitro* and *in vivo*, are needed to elucidate the exact mechanism of heavy metals on male reproductive functioning at the molecular level. *Iran. Biomed. J. 15 (1 & 2): 38-43, 2011*

**Keywords:** Sperm, Creatine kinase, Infertility, Heavy metals

**INTRODUCTION**

Heavy metals are well known environmental and industrial pollutants. Recent studies have shown a considerable increase in metal contamination in relation to the worldwide distribution and extensive use of these chemical agents [1]. Unlike organic pollutants, metals cannot be degraded easily and accumulate throughout the food chain, producing potential human health risks and ecological disturbances [2]. Animal experiments and epidemiological studies have shown that heavy metals are reproductive toxicants [3]. Toxicological studies have demonstrated that these environment pollutants can accumulate in testes and/or epididymis impairing their endocrine and reproductive function [4, 5]. Heavy metals also seem to have a direct effect on sperm cells by reducing their motility and/or affecting their morphology. These adverse effects have been reported either by epidemiological studies of occupationally exposed individuals [6] or animals [7] and *in vitro* studies [8], highlighting a positive correlation with high concentrations of metals in semen.

Creatine kinase (EC.2.7.3.2) is widely distributed in tissues which require high energy. It reversibly catalyzes the phosphorylation of creatine with ATP (creatine + ATP ↔ creatine-phosphate + ADP + H\(^+\)). Its biological role is to provide an ATP buffering system for tissues which require large amounts of energy [9]. Sperm moves using ATP and phosphoryl creatine shuttle as energy sources [10]. Therefore, proper functioning of creatine kinase is a main factor of energy preparation for sperm movement.

Infertility is common in couples of childbearing age. Approximately half of the infertility problems are related to a male factor [11]. According to the
previous reports, exposure to heavy metals can affect the male reproductive system [3-6, 12]. In recent years, the amount of heavy metals is widely increased in the environment. Rapid industrialization, increase in human population, city traffics and increased use of diesel generators/diesel exhaust are believed to be responsible for the increased release of toxic metals into the environment. Exposure to these metals occurs through diet, air, drinking polluted water and ingestion of dust [12, 13]. While a great deal of research has been conducted on the toxic effects of heavy metals on male reproductive function [3, 7, 8, 12, 13], the literature concerning the effects of heavy metals on sperm creatine kinase activity is limited. Thus, the present study was conducted to compare the sperm creatine kinase activity is limited. Thus, the present study was conducted to compare the sperm creatine kinase activity in a dose-dependent manner, and the most effect of inhibition was at 60 µg ml⁻¹. According to the Figure 1, these metals are significantly able to inhibit human sperm creatine kinase activity by approximately 70%, 68%, 66.5%, 65.7%, 64.7% and 62.4%, respectively (Fig.

**MATERIALS AND METHODS**

**Materials.** ADP, AMP, NADP, glucose 6-phosphate dehydrogenase, hexokinase, Triton X-100 and DEAE cellulose were obtained from Sigma Chemical Co. (St. Louis, MO, USA). EDTA, Pb²⁺, Ag²⁺, Sn²⁺, Hg²⁺, In³⁺, Bi³⁺ and BSA were obtained from Merck Chemical Co. (Darmstadt, Germany). All other reagents used were of the highest grade and purity available. All metals as chloride salts were dissolved in double distilled-deionized water.

**Methods.** The semen sample collection and creatine kinase isolation from human sperm were performed according to our previous study [14]. Briefly, the semen samples were analyzed based on WHO [15]. Only samples with the following seminal characteristics were included in the study: volume ≥ 3.0 ml; sperm concentration / ml ≥ 50 x 10⁶; forward motility ≥ 60%; atypical forms ≤ 40%. Creatine kinase was isolated from the samples, as we previously described [14]. Chromatography fractions were collected and monitored for protein content at 280 nm. In addition, an aliquot form each sample was removed for determination of total protein [16], creatine kinase activity [17] and electrophoresis on SDS-PAGE [18].

Creatine kinase activity in different steps of isolation was measured by Rosalki method [17]. As we mentioned previously [14], this method is based on the ability of creatine kinase to reduce NADP in presence of glucose, hexokinase and glucose 6-phosphate dehydrogenase.

The effects of Pb²⁺, Ag²⁺, Sn²⁺, Hg²⁺, In³⁺, Bi³⁺ chloride on the creatine kinase activity were examined by incubation of 1 unit of creatine kinase (10 µg ml⁻¹) with different concentrations (0-100 µg ml⁻¹) of these metals in Tris buffer (0.1 M, pH 6.8) at 25°C for 10 minutes. Degree of inhibition of creatine kinase activity by different concentrations of Pb²⁺, Ag²⁺, Sn²⁺, Hg²⁺, In³⁺ and Bi³⁺ was determined at 340 nm as previously described [14, 17]. Then, inhibition kinetic was determined by incubation of 1 unit of creatine kinase with 0.3 mM Pb²⁺, 0.6 mM Ag²⁺, 0.5 mM Sn²⁺, 0.3 mM Hg²⁺, 0.5 mM In³⁺ and/or 0.3 mM Bi³⁺ in presence of concentrations of 0 to 10 mM creatine phosphate (as creatine kinase substrate).

**Statistical analysis.** Results are presented as mean ± standard deviation (SD). All assays were performed in triplicate and the mean was used for the calculation. Creatine kinase activity in absence (as control) and/or presence of heavy metals was compared using analysis of variance (ANOVA). The statistical significance was accepted when p<0.05. All analyses were carried out using the Statistical Package for the Social Sciences (SPSS) 15.0 software.

**RESULTS**

Creatine kinase isolation from human sperm was confirmed according to our previous study [14]. To study in vitro effects of metal compounds on human sperm creatine kinase activity we first examined the effects of six metals ions (In³⁺, Bi³⁺, Sn²⁺, Ag²⁺, Pb²⁺ and Hg²⁺) on activity of this enzyme at six different concentrations: 5, 20, 40, 60, 80 and 100 µg ml⁻¹ (Fig. 1). According to the Figure 1, these metals are significantly able to inhibit human sperm creatine kinase activity in a dose-dependent manner, and the most effect of inhibition was at 60 µg ml⁻¹ concentration. Although all these metals with concentrations of 5 to 100 µg ml⁻¹ in comparison to the control (metal-free) showed almost similar inhibition effect on creatine kinase activity, Hg²⁺ had the most powerful effect (Fig. 1). This study showed that 60 µg ml⁻¹ concentrations of Hg²⁺, Pb²⁺, Ag²⁺, Sn²⁺, Bi³⁺, and In³⁺ can inhibit human sperm creatine kinase activity by approximately 70%, 68%, 66.5%, 65.7%, 64.7% and 62.4%, respectively (Fig.
Ag, Pb, 80
Hg, 60
In, Bi, 65.7
68
Sn, 100
40
20
Hg, Bi, 66.5
Sn, Pb

sperm creatine kinase activity as
Therefore, these heavy metals can inhibit human
under these assay conditions (Fig. 3, lower panel).

Gennart et al. [20] found when
workers were exposed to high levels of Pb\(^{2+}\), their
fertility decreased. Xu et al. [21] reported a
significant correlation between cadmium (Cd\(^{2+}\)) and
sperm concentration. Lerda [22] reported that a
decrease in sperm concentration occurred in men
with a mean blood Pb\(^{2+}\) concentration of 48.6 μg dl\(^{-1}\)
(about 0.0023 mM). The addition of heavy metals to
human semen can reduce sperm motility [23, 24].
Young et al. [25] observed that hyper-activation of
rabbit sperm was inhibited by heavy metals.
Although it is known that heavy metals are
associated with an overall reduction in men fertility,
the mechanism is not clear yet.

Creatine kinase is an important enzyme which
plays a major role in sperm energy homeostasis [11].
Thus, in the present study effects of some heavy
metals which are more present in the environment
such as Hg\(^{2+}\), Pb\(^{2+}\), Ag\(^{+}\), Sn\(^{2+}\), Bi\(^{3+}\) and In\(^{3+}\) were
evaluated on human sperm creatine kinase activity in
an in vitro model. After isolation of creatine kinase
from human sperm, the experiments were conducted
with various concentrations of these heavy metals
(0-100 μg ml\(^{-1}\)) on activity of this enzyme, as a
necessary key enzyme for sperm normal
metabolism. At first, it was observed which all these
metals significantly \(P<0.01\) inhibited creatine
kinase activity in a dose-dependent manner.

Maximum inhibition was seen with Hg\(^{2+}\), followed
by Pb\(^{2+}\) > Ag\(^{+}\) > Sn\(^{2+}\) > Bi\(^{3+}\) > In\(^{3+}\). Then, the kinetic
analysis of the effect of these metals on human
sperm creatine kinase activity showed that all heavy
metals in this study could act as competitive
inhibitors of human sperm creatine kinase. This
study also showed maximum \(K_i\) for Hg\(^{2+}\) is 8.34 mM
and then followed by Pb\(^{2+}\) (5 mM) > Ag\(^{+}\) (4.54
mM) > Sn\(^{2+}\) (3.45 mM) > Bi\(^{3+}\) (3.12 mM) > In\(^{3+}\)
(2.63 mM), which confirm results of the inhibition

**DISCUSSION**

Heavy metals are widely distributed in the
environment and these metal ions are present in low
concentrations in the semen of healthy and
presumably unexposed men [19]. Men working in
industries with high exposure to heavy metals have
been reported to have elevated levels of these metals
in their blood [20]. Gennart et al. [20] found when

![Fig. 1. In vitro effects of lead, silver, tin, mercury, indium and bismuth on creatine kinase activity obtained from human sperm. Samples (10 μg ml\(^{-1}\) creatine kinase equal with 1 unit activity) were incubated with different metal concentrations (5-100 μg ml\(^{-1}\)) for 10 min at 25°C. Results are expressed as percentage of the corresponding control activities (unit per liter). Data have represented as the mean ± S.D. of triplicate determinations.](http://IBJ.pasteur.ac.ir)

![Fig. 2. The inhibition percentage of human sperm creatine kinase activity in presence of 60 μg ml\(^{-1}\) indium, bismuth, tin, silver, lead and mercury compared with control (in absence of heavy metals).](http://IBJ.pasteur.ac.ir)
Fig. 3. (Upper panel) Heavy metals-mediated inhibition of human sperm creatine kinase in a varying concentrations of substrate (0-10 mM). Concentrations of 0.3 mM of bismuth (Bi\(^{3+}\)), 0.3 mM of lead (Pb\(^{2+}\)), 0.3 mM of mercury (Hg\(^{2+}\)), 0.5 mM of indium (In\(^{3+}\)), 0.5 mM of tin (Sn\(^{2+}\)) and 0.6 mM of silver (Ag\(^{2+}\)) were used. Results are expressed as mean ± S.D. of triplicate determination. (Lower panel) Lineweaver-Burk plots reflect a probable competitive type inhibition of human sperm creatine kinase activity by heavy metals.

percentage of human sperm creatine kinase activity in the presence of each one of metals above. Previous studies by others suggested that creatine kinase possess two sites, one for the nucleotide (Mg-ATP) and the other for the guanidine substrate (phosphocreatine) [26, 27]. These studies also indicated that the most likely mechanism for phosphorylation of ADP by phosphocreatine involves a highly reactive metaphosphate intermediate that is generated in the rate-determination step by proton transfer and cleavage of the P-N bound of phosphocreatine [27, 28]. The most important role which creatine kinase plays in catalysis of this reaction involves protonation of the P-N nitrogen of phosphocreatine.

There is evidence that a sulfhydryl group (cysteine residues) and an imidazole group (histidine residues) are involved in catalysis at the active site of creatine kinase. Either the -SH group or imidazolium group could function as a general acid catalyst for protonation of the P-N nitrogen [28]. According to these reports, presence of Mg\(^{2+}\) and -SH group at creatine kinase active site is necessary for its activity [26-28]. Different reports have shown that heavy metals with different mechanisms can inhibit enzymes. Mercuric chloride and sodium selenite can oxidize -SH groups of different enzymes and thus Pb\(^{2+}\) to the inhibition of a large number of sulphydryl-containing enzymes [29]. Farina et al. [29] indicated that Hg\(^{2+}\) inhibits the activity of δ-aminolevulinate dehydrogenase in mouse liver, kidney and brain by oxidizing -SH groups located at the active center of the enzyme. Shinozaki and Pritzker [30] and Waalkes [31] reported that Pb\(^{2+}\) and cadmium can inhibit the alkaline phosphatase activity in brain, through Zn\(^{2+}\) substitutions.

Our previous study showed cadmium as an important heavy metal in tobacco probably is able to inhibit activity of this enzyme via displace magnesium ions in human sperm creatine kinase [14]. The results of this study are also in agreement with other findings. We indicated that although human sperm creatine kinase is inhibited by all these metals in approximately the same range, divalent cations, especially, mercury have the most effectiveness on human sperm creatine kinase activity. As mentioned above, mercury ions via SH group oxidation at active site of different enzymes can inhibit their activity. This can be an important reason for more inhibitory effect of mercury compared with other ions in this study. Therefore, inhibition effects of these metals on activity of human sperm creatine kinase may be due to –SH group oxidation at active site or displacement of Mg\(^{2+}\) and/or both.

In conclusion, our results indicated that Hg\(^{2+}\), Pb\(^{2+}\), Ag\(^{2+}\), Sn\(^{2+}\), Bi\(^{3+}\) and In\(^{3+}\) reduce in vitro creatine kinase activity in human sperm, by an apparently competitive inhibition, possibly through displacement of Mg\(^{2+}\) in this enzyme. Creatine kinase has an important role in sperm energy homeostasis [9, 10]. Therefore, diminution of creatine kinase activity by these heavy metals may potentially impair sperm functional parameters. However, further in vitro and in vivo investigations are required to confirm the role played by these heavy metals in male infertility.
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REFERENCES


