Atorvastatin-Infused Carboxymethyl Cellulose-Gelatin Hydrogel: A Synergistic Strategy for Enhanced Wound Healing and Skin Tissue Regeneration

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ABSTRACT

Background: Skin tissue engineering is a live we alternative to traditional methods for addressing skin. So This study aimed to synthesize a hydrogel consisting of the detection and wound healing in an animal model.

Methods: Five unique in a tions of hydrogel with different concentrations of ATP (0.14, 0.16, 1%, and 2% w/v) were synthesized using CMC-Gel. The stratural concentrations of the hydrogels were assessed using SEM and FTIN ecthoscopy. Additional evaluations included swelling behavior, deladability ATR release, compatibility, hemolytic activity, and the toxic entropy on the viability of NIH/3T3 fibroblast cells. Their therapeutic effective less on the hydrogels in enhancing wound healing was investigated in an animal model by making a full-thickness skin incision in What its.

Res is: The synthesized CMC-Gel scaffolds had a porous structure with interconnected pores measuring $103\pm8.74~\mu m$, with the ability to enhance cell migration. The MTT analysis showed a dose-dependent relationship between ATR and cell proliferation, and its desirable concentration was 0.1% w/v. Furthermore, increased ATR concentrations were associated with decreased dressing capacity for hemostasis and coagulation. In vivo studies revealed that all the hydrogel-treated groups significantly outperformed the gas-treated control group in promoting wound closure rates. Remarkably, the CMC-Gel-ATR 0.1% group exhibited the highest rates of wound closure, re-epithelialization, and angiogenesis.

Conclusion: Our results demonstrated that CMC-Gel-ATR may be proposed as a desirable wound dressing for clinical application due to its unique physicochemical properties and comprehensive biocompatibility in in vitro and in vivo investigations. *DOI:* 10.61186/ibj.5031

Keywords: Angiogenesis, Atorvastatin, Biopolymers

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List of Abbreviations:

ATR: atorvastatin; BCI: blood clotting index; CMC: carboxymethyl cellulose; FBS: fetal bovine serum; FTIR: fourier-transform infrared spectroscopy; GeI: gelatin; H & E: hematoxylin and eosin; MTT: 3'-(4,5-Dimethylthiazol-2-yl)-2,5-diphenyl tetrazolium bromide; PBS: phosphate-buffered saline; SEM: scanning electron microscopy

INTRODUCTION

kin injuries are among the most common medical issues, often resulting in high medical costs, reduced individual productivity, and limited physical activities^[1]. These injuries disrupt the normal function of the skin, causing the body to lose essential fluids and proteins, alter metabolic processes, and affect the circulatory and defense systems^[2].

Traditionally, wound treatment strategies included natural substances such as honey and plant extracts. Today, numerous types of dressings have been developed to accelerate wound healing. One of the most significant advancements in dressing production is the use of polymeric materials. Among them, hydrogels have a special capability because of their three-dimensional network structure. Additionally, due to their fluid-absorption capacity, these materials can efficiently absorb exudate from wounds and making a moist wound environment^[3].

CMC and Gel are two natural and semi-synthetic polymers that can be used to create biomedical hydrogels. **CMC** features attractive surface characteristics that enhance protein adsorption, facili cell aggregation and differentiation, and provide lo degradation rates, high mechanical streng and excellent biocompatibility. Furthermore, can retain moisture in wound areas, where omo the growth of the extracellular matrix and regeneration of the epithelium^[4]. Gel, another flowing tible polymer produced by hydrolyzing co. ven. rovides several advantages including advantages, including osafet, ompatibility with biological systems, non-cicit and cost-effectiveness. Its amino acid sequences, such as RGD, enhance hemostasis and stimulate cell adhesion and proliferation throughout the healing process^[5].

ATR is a member of the second generation of statins. In addition to the anti-inflammatory properties, statins are a viable option for wound healing as they can control cellular processes^[6]. Studies have explored the effectiveness of statins in treating wounds and skin disorders using different formulations administration methods, including intraperitoneal injection and oral intake. However, like other systemic medications, oral statins have side effects, including myopathy and liver problems^[7]. Topical drug delivery approaches are suggested as a feasible substitute for oral statins in the wound healing process. This approach can minimize side effects, improve therapeutic effects, and enhance medication delivery^[8].

The present study aimed to exhibit that the incorporation of ATR into a CMC-Gel hydrogel could enhance wound healing by modulating the wound microenvironment and improving tissue regeneration.

While the therapeutic potential of ATR has been well established, its integration into biopolymer-based topical scaffolds remains underexplored. hypothesize that embedding ATR in a CMC-Gel scaffold can promote wound healing by enhancing moisture retention, supporting cell viability and proliferation, and facilitating tissue regeneration. This approach addresses the limitations of conventional treatments and contributes to the development of multifunctional, bioactive wound care systems. To our knowledge, the combination of CMC, Gel, and ATR in a topical hydrogel formulation has not been investigated, representing extensively advancement in hydrogel-based wound healing approaches.

MAY SAND METHODS

Mater

MA CA and Gel powder (bovine skin, type B) we hased from Sigma-Aldrich (USA). Trypsin-TDT DMEM, FBS, and penicillin-streptomycin were sted by Gibco (Germany). ATR was obtained from aana Pharmaceutical Company (Tabriz, Iran). The Pasteur Institute of Iran in Tehran provided the fibroblast cell line NIH/3T3.

Synthesis of CMC-Gel and CMC-Gel-ATR Hydrogels

CMC powder (4% w/v) and Gel powder (3% w/v) were dissolved in deionized water (8:1 ratio) under stirring at 400 rpm for 12 hours at room temperature. ATR was then added and homogenized for six hours. Afterwards, phenoxyethanol was added to 0.9% of the final volume of the hydrogel to enhance stability and provide antiseptic properties (Table 1). The hydrogels were preserved at -80 °C for 16 hours and then transferred to a freeze dryer (Terrace, Spain) set at -54 °C for 48 hours, which allowed a porous structure to be formed^[9].

Characterization of the scaffolds Scanning electron microscopy

SEM was used to evaluate the surface properties of the generated lyophilized hydrogels (SEM; DSM 960A, Zeiss, Germany). To improve conductivity, a small layer of gold was applied to the freeze-dried hydrogels (SCD 004, Balzers, Germany). Image J and Origin Pro software were used to determine the average pore diameter. For this assessment, 20 of randomly chosen places within each image were scanned^[10].

Table 1. Formulation of CMC-Get-ATR for 1 mL of hydroget							
Hadasal	Ratio	Polymer	AT				
Hydrogel	CMC:Gel	Concentration	Concen				

		J		
Hydrogel (sample)	Ratio CMC:Gel (% v/v)	Polymer Concentration (% w/v)	ATR Concentration (mg)	Phenoxyethanol (% v/v)
CMC-Gel	8:1	4:3	0	0.9
CMC-Gel-ATR 1 (0.1% w/v)	8:1	4:3	1	0.9
CMC-Gel-ATR 5 (0.5% w/v)	8:1	4:3	5	0.9
CMC-Gel-ATR 10 (1% w/v)	8:1	4:3	10	0.9
CMC-Gel-ATR 20 (2% w/v)	8:1	4:3	20	0.9

Fourier-transform infrared spectroscopy

FTIR was utilized to examine the interactions among functional groups in different hydrogel structures, particularly CMC-Gel and CMC-Gel-ATR. The FTIR spectroscopy technique (Spectrum GX, PerkinElmer, USA) was used to record the FTIR spectra spanning a wavelength range of 400 to 4000 cm^{-1[11]}.

Evaluation of hydrogel degradation

The rate of degradation was assessed by measuring the weight loss of the hydrogel samples. Before submerging the dried hydrogels in a PBS solution (pH 7.4), we recorded their weights (W₀). The samples were then immersed in the PBS solution for predetermined intervals of 2, 6, 12, 24, and 48 hours. After each interval, excess solution was removed from the samples before reweighing them (W_1) . The percentage of weight reduction was finally calculated using the formula give in Equation $(1)^{[12]}$.

Weight loss (%) =
$$\frac{\text{W0} - \text{W1}}{\text{W0}} \times 100$$

Evaluation of hydrogel water up

The ability of hydrogel samp (t) bsort water was evaluated at room-temp A freeze by hydrogels. The (W₂) were A freeze-drying procedure was used to d samples (W₀) were initial weights of the ir ir persion in PBS for varying recorded before ti \mathcal{N}_{2} periods (3, 6, 12, and 48 hours). The samples were reweighed (W₁) at each specified time interval, and equation (2) was used to determine the water absorption^[13].

Water absorption (%) =
$$\frac{W1 - W0}{W0} \times 100$$
 (2)

Evaluation of hydrogel blood compatibility

A 2 mL volume of rat blood was diluted with 2.5 mL of normal saline. The diluted blood sample (200 µl) was then added to three identical hydrogel samples (100 µL) prepared in 96-well plates. After 1 hour of incubation at 37 °C, the plates were centrifuged (1500 rpm, 10 minutes,). For the experimental controls, a negative control was prepared by combining 4 mL of normal saline with 200 µL of diluted blood, while a positive control was prepared by mixing 4 mL of distilled H₂O with 200 µL of diluted blood^[14]. The supernatant was then tested for absorbance at 545 nm, and the percentage of hemolysis was determined using Equation (3).

Hemolysis (%) =
$$\frac{Ds - Dnc}{Dpc - Dnc} \times 100$$
 (3)

Evaluation of hydroll's BCI

Fresh rat bloc worklitted in a citrated tube to measure the two cop lation index of the samples.

Next, 1.5 of the sample was incubated at 37 °C for lext, 1.5 compared was incubated at 37 °C for how the unpressive then combined with 100 μL cool of rats. The mixture was incubated (5 inut) before adding 0.2 M of CaCl₂. After an Vitional five minutes, each sample was mixed with 25 of distilled H₂O. The resultant suspensions were ansferred to a 96-well plate for further analysis following a final incubation of five minutes. To create the control samples, we combined 25 mL of deionized H₂O with 100 µL of fresh human blood^[15]. Using equation (4), we measured the relative absorbance of the hydrogel samples at 545 nm to calculate the BCI.

$$BCI (\%) = \frac{A \text{ sample}}{A \text{ control}} \times 100 \tag{4}$$

Evaluation of hydrogel's cytotoxicity effects

The effect of CMC/Gel and CMC/Gel/ATR hydrogel dressing extracts on NIH/3T3 fibroblast cell line was carried out by using the MTT assay. The NIH/3T3 fibroblast cell line were seeded at a density of 1×10^4 cells per well in a sterile 96-well plate containing DMEM supplemented with 10% FBS. The cells were then incubated with 5% CO₂ at 37 °C overnight. To prepare hydrogel extracts, we incubated hydrogel samples containing different concentrations of ATR with DMEM for 48 hours under sterile conditions. After a 24-hour incubation period, 100 µL of the extracts were added to each well in triplicate. Following this incubation period, 20 µL of MTT solution (5 mg/mL⁻¹)

was added to each well, and the supernatants were discarded. After three hours of further incubation at 37 °C, 100 µL of DMSO was added to dissolve the formazan crystals^[16]. The absorbance at 570 nm was then measured using a plate reader to determine the percentage of cell viability, which was calculated using equation 5.

Cell viability (%) =
$$\frac{\text{absorbance of sample}}{\text{absorbance of control}} \times 100$$
 (5)

Evaluation of release of ATR

To determine the amount of ATR released from the CMC-Gel hydrogel, a standard curve must be created for ATR using a range of 0.1 to 100 mg./mL⁻¹. CMC-Gel-ATR 1 mg/mL⁻¹ hydrogel was placed in 5 mL of simulated body fluid within a shaker incubator kept at 37 °C (50 rpm). The absorbance of the supernatants was measured at 246 nm at different time intervals: 2, 4, 6, 12, 24, 36, 48, 72, and 96 hours, to evaluate the release of ATR^[17]. Finally, by analyzing the established standard curve for ATR, we transformed the spectroscopic results into quantitative values.

Evaluation of hydrogel's pH variation

At predetermined intervals of 2, 4, 6, 12, 24, and hours, the pH of the normal saline contain ng the hydrogel samples was measured at room using a digital pH meter (pH/mV Met monitor any potential pH variations ced by the hydrogel^[18].

In vivo experiment grows

Adults male Wistar rate, weighing 200–250 g)

based from Share University of Medical

University of Medical Sciences (Shahroud, Iran). The animals were provided food and water in standard housing conditions. The produced hydrogels were evaluated for their effectiveness in wound healing using a full-thickness wound model (1.5 \times 1.5 cm²). Rats were given an intraperitoneal injection of 100 mg/kg⁻¹ of ketamine and 10 mg/kg⁻¹ of xylazine to induce anesthesia. The dorsal skin of rats was cleaned, shaved, and disinfected using 70% alcohol. Afterwards, the rats were randomly divided into three groups of six. Group 1: negative control (treated with sterile gauze); Group 2: treated with CMC-Gel hydrogel; Group 3: treated with CMC-Gel-ATR hydrogel. The animals were sacrificed 14 days after the procedure, and the wound tissue was collected for histological analysis. Photographs taken 14 days following surgery were used to monitor the wound healing process^[19]. Image J software was used to evaluate the results. Equation 6 was employed to calculate the percentages of wound closure.

Wound closure (%) =
$$1 - \frac{\text{wound area}}{\text{initial area}} \times 100$$
 (6)

Histopathological analysis

The development of granulation tissue, angiogenesis, formation, collagen deposition and reepithelialization were assessed using Masson's trichrome and H&E staining. Tissue specimens were preserved in 10% formalin for histological examination. After being dehydrated and fixed in paraffin, the specimens were cut into 5 µm-thick sections^[20]. We used a light microscope (Olympus BX51; Olympus, Tokyo, Japan) to analyze the histology slides.

Statistical analysis

GraphPad Prism 10.2 Software was used for all statistical analyses. For oreal distributed data, a one-way analysis of va NOVA), followed by Tukey's post h was used for multiple group on-normally distributed data, we compariso Wallis test, followed by the Mannst. Ďata are presented as mean \pm SD. A p05 was considered statistically significant.

RESULTS

Morphological properties of hydrogels

The SEM analysis of the CMC-Gel scaffold (Fig. 1A-1C) revealed a porous architecture characterized by larger pores (103 \pm 8.74 μ m), which may not be ideal for promoting cell adhesion and migration. In contrast, the CMC-Gel-ATR20 scaffold (Fig. 1D-1F) exhibited a finer porosity and a more consistent fibrous network (62.39 ± 5.07) , offering an increased surface area for cell attachment. Recent research indicates that this characteristic is crucial for enhancing the regeneration of damaged tissues. Therefore, incorporating ATR into the CMC-Gel-ATR scaffold has improved its functional attributes.

FTIR spectral characteristics

Figure 1G and 1H show the results of FTIR spectra of CMC-Gel and CMC-Gel-ATR20 scaffolds. A broad band at 3400-3300 cm⁻¹ corresponds to O-H and N-H stretching in CMC and Gel, respectively. The peak near 2900 cm⁻¹ is from CMC and amide I from Gel, while 1550 cm⁻¹ corresponds to the amide II band. The peak at 1420 cm⁻¹ reflects symmetric COO⁻ stretching, and the range between 1050 and 1100 cm⁻¹ relates to C-O-C and C-O stretching. Shifts in these bands suggest hydrogen bonding between CMC and Gel, confirming their compatibility (Fig. 1G). For CMC-Gel-ATR20 (Fig. 1H), the peaks at 1695.04 and 1714.72 cm⁻¹ indicate C=O stretching associated with ATR.

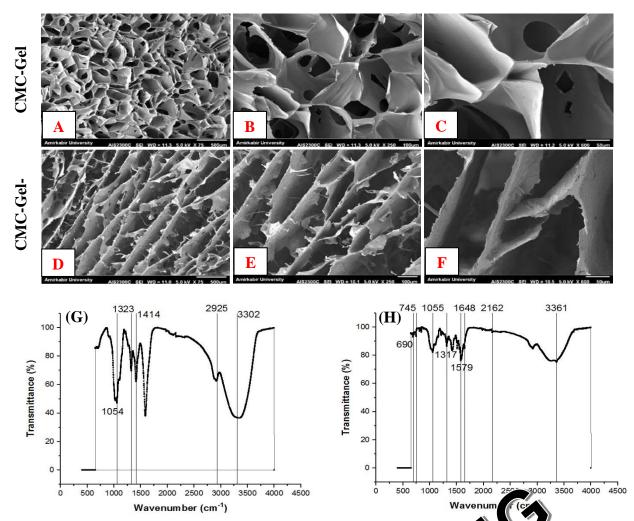


Fig 1. SEM of CMC-Gel (A, B, C) and , CMC-Gel-ATR (D, E, F) with 75×, 250×, and 500× magnif. tid. FT spectrums of CMC-Gel (G) and CMC-Gel-ATR20 (H) at 400 to 4000 cm⁻¹.

Additional peaks at 1648.21 and 2162.04 cm⁻¹ likely correspond to conjugated C=C vibrations in aromatic rings. Bands at 1579.07, 1591.27, and 1517.63 cm⁻¹ further support the presence of aromatic structures. The 1404 to 1424 cm⁻¹ region relates to C–H bending or ester C–O stretching, and the 689.96 cm⁻¹ band confirms the presence of mono-substituted benzene rings. These findings affirm the successful incorporation of

Degradation behavior of hydrogels

The weight loss of the scaffold was ssed at intervals of 2, 6, 12, 24, and 48 hg (s. CM \sim 1 showed the highest degradation (98.0, 3.2%) at 48 h, followed by CMC-Gel-ATR1 (94.5 \pm 5.36%), ATR5 (93.5 \pm 2.12%), ATR20 (88.5 \pm 2.13%), and ATR10 (87 \pm 4.24%). After 2 hours, the degradation rates were as follows: 22 \pm 2.82% (CMC-Gel), 19 \pm 1.41% (CMC-Gel-ATR1), 13.5 \pm 2.12% (ATR5), 19.5 \pm 0.78%

(ATR20). and (2.5) 0.79% (ATR10). Although the water ison in of ATR may influence scaffold decoda in a higher doses, no statistically significant are like ere observed among the treatment groups at (p) 20.05). All comparisons were made related to the CMC-Gel group.

Swelling behavior of hydrogels

All groups demonstrated a time-dependent increase in water uptake from 3 to 48 hours. CMC-Gel exhibited the swelling rates, reaching $16.5 \pm 2.12\%$ $85.22 \pm 10.21\%$, and $109.01 \pm 12.94\%$ at 3, 24, and 48 hours, respectively. Although ATR-containing hydrogels showed a similar swelling trend over time, a dose-dependent reduction in water absorption was observed. Following 48 hours, CMC-Gel-ATR1, ATR5, ATR10, and ATR20 reached $108 \pm 10.53\%$, $96.5 \pm 7.26\%$, $94 \pm 6.05\%$, and $77.5 \pm 3.92\%$,

respectively. Notably, the ATR20 showed a significant reduction in water uptake compared to CMC-Gel at both 24 hours and 48 hours ($67 \pm 2.82\%$ vs. $85.22 \pm 10.21\%$) (p < 0.05), while other groups did not exhibit any significant difference.

Hemolysis behavior of hydrogels

The hemolysis rate in the CMC-Gel was $4.71 \pm 1.39\%$. The ATR-containing groups showed a dose-dependent hemolysis rates as follows: ATR1 $(2.5 \pm 1.3\%)$, ATR5 $(3.57 \pm 1.35\%)$, ATR10 $(3.81 \pm 1.25\%)$, and ATR20 $(5.47 \pm 1.49\%)$. All treatment groups differed significantly from the negative control group, with a hemolysis rate of $0.2 \pm 0.1\%$ (p < 0.05). In contrast, the positive control showed a highly significant increase in hemolysis rate, $99.33 \pm 5.37\%$ (p < 0.0001). An increase in the ATR dosage resulted in the heightened hemolysis, implying that only ATR20 exhibited potential hemolytic effects.

BCI of hydrogels

The BCI value of the CMC-Gel scaffold was found to be $27.31 \pm 4.15\%$. A dose-dependent increase was observed in BCI at high concentrations of ATR. The BCI values for CMC-Gel-ATR1, ATR5, and ATR10 were $29.06 \pm 3.3\%$, $33.57 \pm 3.84\%$, and 37.84 ± 4.25 respectively, with no statistically significant differen compared to CMC-Gel (p > 0.05). However, the CMC Gel-ATR20 demonstrated a significant increase in BCI $(44.39 \pm 6.70\%; p < 0.01)$, indicating otting ced he positive potential at higher ATR concentrations 5.39%, which control showed a BCI value of was highly significant differen from t ≥ **C**MC-Gel (*p* < (0.0001). This outcome effectiveness of ATR as a wound dressin which is highly dependent on dosage, i.e. at high concentrations, it may not be effective for managing bleeding wounds.

Analysis of MTT assay results

The cytotoxic effects of the prepared hydrogels on the cell viability are presented in Figure 2. These results indicate that low levels of ATR can maintain and even improve the fibroblast viability, while higher ATR concentrations may impart cytotoxicity. Therefore, 0.1% ATR was chosen for the animal study.

ATR release analysis

The cumulative release profile of ATR over various time intervals is illustrated in Figure 3. The amount of the released ATR was quantified by constructing a standard release curve. After 12 and 24 hours, the recorded releases of A from the hydrogel were 614.91 \pm 22.70 μg and 80 25.41 μg , respectively, leading to a total reval of 9.9.61 \pm 14.33 μg over a four-day dural 11, see results suggest a gradual release of A1. In the CMC-Gel hydrogel, reaching a platear case a er 6 hours (949.61 \pm 33.24 μg). No addit anal Nease was observed at the 72 and 96-hour

nal sis of hydrogel pH variation

Table 2 reports the pH values of the synthesized hydrogels over 48 hours. All groups, except for CMC-Gel (7.13 ± 0.40), were in the slightly acidic pH range (6.22-6.81).

In vivo wound healing findings

The study evaluated the efficacy of CMC-Gel and CMC-Gel-ATR1 hydrogels using a full-thickness wound model (Fig. 4A). The negative control, which its

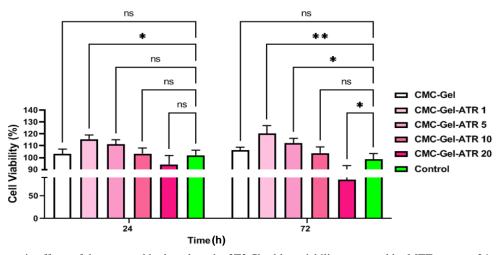


Fig 2. The cytotoxic effects of the prepared hydrogel on the 3T3 fibroblast viability measured by MTT assay at 24 and 72 h. Values represent the mean \pm SD; n = 6; ns: not significant; *p < 0.05, and **p < 0.01.

Mousavi et al. Wound Healing by ATR

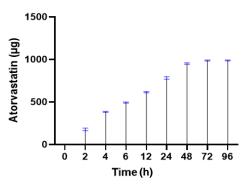


Fig 3. The release of ATR from CMC-gel hydrogel in PBS solution measured after 96 hours in a shaker incubator at 50 rpm at 37 °Cr. Values represent the mean \pm SD (n = 3).

wounds had been covered with sterile gauze, exhibited signs of infection and inflammation, leading to incomplete healing after two weeks. In contrast, treated groups demonstrated significant healing during the same timeframe. Macroscopic photographic evaluations indicated that one-week post-treatment, the hydrogel groups experienced minimal bleeding. Notably, the CMC-Gel-ATR1 yielded the most promising outcome showing no indications of infection or inflamma Figure 4B presents a quantitative analysis of healing based on the wound closure rates. The rates for the negative control, CMC-Gel, a ATR1 groups on days 7 and 14 we ecorded as follows: 28.34 ± 10.66%, 70.2 4.28 39.42 ± 14.39%, $87.66 \pm 5.48\%$, $52.76 \pm$ 1%, and 93.05 \pm 6.17%, respectively. The significant improvement it wour ATR1 showed a healing compared to both the control an CMCgroups. Both CMC-Gel and CMC-Gel-A7 g ups displayed significant improvements compared to the control group on days 7 and 14; however, the effect was more pronounced in the CMC-Gel-ATR1 group (p < 0.01 vs. p < 0.05). This observation highlights the considerable potential of the CMC-Gel-ATR1 hydrogel in reducing healing time.

Histopathological findings

All treatment groups (negative control, CMC-Gel, and CMC-Gel-ATR1) demonstrated the formation of an distinct epidermal layer after 14 days. The CMC-Gel-ATR1 group showed thicker re-epithelialization and

more separation of the dermal layer (Fig. 5L). The CMC-Gel formed a crusty scab, which is a negative aspect of the healing process, as it was absent in the CMC-Gel-ATR1 group Fig. 5H). Blood vessel formation was more pronounced in both the negative control and CMC-Gel-ATR1 groups compared to the CMC-Gel group. However, many blood vessels in the negative control remained immature (Fig. 5C and 5L). Notably, only CMC-Gel-ATR1 group exhibited the development of skin appendages and resulted in a greater thickness of the dermis layer of the skin (Fig. 5J). The CMC-Gel-ATR1 was particularly successful in forming hair follicles (Fig. 5K). Additionally, an increased number of sebaceous glands was observed (Fig. 5K). Furthermore, collagen fiber synthesis was highest in CMC-Gal ATR1 group (Fig. 5K). The usso trichrome study of all groups findings from the 1 formation of collagen fibers are shown in K group (Fig. 6K) was organized in the CM to the other groups.

DISCUSSION

Wound healing is a dynamic, multi-phase process. It begins with hemostasis, followed by the inflammatory and the proliferative phases, which aims to close the wound. During the remodeling phase, collagen fibers regenerate. Scaffolds made from CMC and Gel are for their biocompatibility, mechanical flexibility, and ability to replicate the extracellular matrix^[21]. In a previous study, Soleimani et al. have investigated a hydrogel composed of CMC gel, revealing that it has adequate mechanical strength and antibacterial properties. Their study also verified that the hydrogel is non-toxic and plays a significant role in wound healing^[22]. Saghafi et al. have demonstrated that topical application of the gel and nanogels containing ATR significantly accelerates wound healing and reduces pain without notable side effects, indicating ATR potential for wound management^[8].

Studies of the effects of ATR on the survival and function of dermal fibroblasts are very limited. In one study, the treated human dermal fibroblasts with 0.5-2.5 μ M of ATR showed no toxic effects after 24 h^[23].

Table 2. pH values of the treated groups

Table 2. pri values of the treated groups								
	Samples	0 h	2 h	4 h	6 h	12 h	24 h	48 h
	CMC-Gel	7.18 ± 0.45	7.15 ± 0.39	7.14 ± 0.41	7.15 ± 0.43	7.14 ± 0.38	7.15 ± 0.41	7.13 ± 0.40
	CMC-Gel-ATR 1	6.93 ± 0.24	6.84 ± 0.30	6.83 ± 0.28	6.83 ± 0.29	6.82 ± 0.31	6.82 ± 0.29	6.81 ± 0.24
	CMC-Gel-ATR 5	6.85 ± 0.25	6.82 ± 0.29	6.82 ± 0.22	6.81 ± 0.26	6.85 ± 0.25	6.82 ± 0.29	6.82 ± 0.22
	CMC-Gel-ATR 10	6.72 ± 0.19	6.70 ± 0.22	6.69 ± 0.20	6.68 ± 0.19	6.85 ± 0.25	6.82 ± 0.29	6.82 ± 0.22
	CMC-Gel-ATR 20	6.35 ± 0.36	6.28 ± 0.25	6.26 ± 0.19	6.25 ± 0.23	6.69 ± 0.15	6.67 ± 0.17	6.66 ± 0.16

Values represent the mean \pm SD (n = 6).

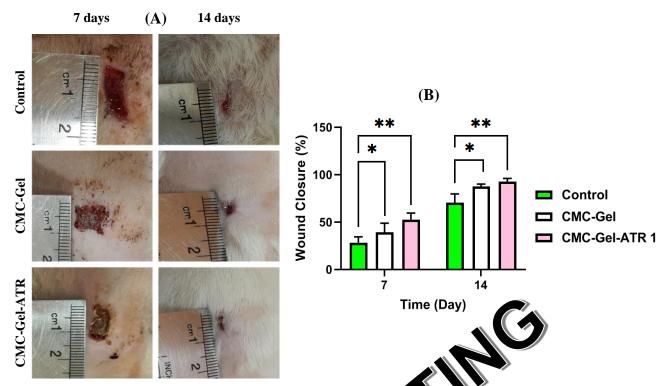


Fig 4. (A) In vivo wound-healing results; a macroscopic appearance of wounds the with solid gauze, CMC-Gel, and CMC-Gel-ATR1 hydrogels at 7 and 14 post-wounding; (B) a histogram comparing the per tages of wound closure among the experimental groups after 7 and 14 days following the initial wound. Values represent the (n = 6); p < 0.05, p < 0.05, p < 0.01.

Another study has reported that concentrations more than 20 µM for 72 h, could significantly decre viability due to oxidative stress and suppressing antioxidant enzymes^[24]. MTT results from support that lower ATR concentrations are while higher doses show cytotoxicity. have suggested that pore sizes between μm are optimal for wound healing because ev facilitate affe is exhibited pore effective cell migration. Our sizes within this optimal range suggesting their potential to support efficient cell migration and nutrient exchange^[24]. Gradual degradation of the scaffolds is crucial to avoid inflammation from residual materials and support cell growth. 26. The scaffolds retained favorable biodegradability characteristics even after drug incorporation. CMC and gel scaffolds absorb and retain moisture due to their hydrophilic and porous structure, aiding chronic wound management. Moisture retention not only prevents wound dryness but also enhances the controlled release of ATR. Likewise, scaffold swelling influences drug permeability and diffusion rate [25].

Water uptake analysis confirmed the ability of hydrogels to maintain a moist environment, which is essential for healing^[22]. The initial interaction between the implanted wound dressing and red blood cells

an inflammatory response. Zhou and Yi have oborated that the acceptable hemolysis threshold for wound dressings is below 5%^[14]. Blood compatibility results of our study showed that all formulations remained below the critical 5% hemolysis threshold, except for ATR20. This finding indicates a dosedependent hemocompatibility, which is essential for clinical applications. Platelet aggregation coagulation are crucial for hemostasis, with a higher BCI suggesting slower clotting and enhanced blood compatibility^[25]. A trial involving hypercholesterolemia patients has found that 24 weeks of ATR administration reduces mean platelet volume and the neutrophil-tolymphocyte ratio, suggesting antiplatelet and antiinflammatory effects of ATR^[26]. Thus, the use of ATR in wound dressings is dose-dependent in which high doses can impair clotting and are unsuitable for treating bleeding wounds.

The range of pH of the wound environment plays a critical role in the healing process. The alkaline pH range (7.15–8.9), which are common in chronic wounds, slow down healing. However, an acidic environment promotes healing. Based on our results, all ATR-loaded samples showed acidic pH, which supports wound healing^[27]. In vivo data corroborated our in vitro findings, showing that ATR at a concentration of 1 µM

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Wound Healing by ATR Mousavi et al.

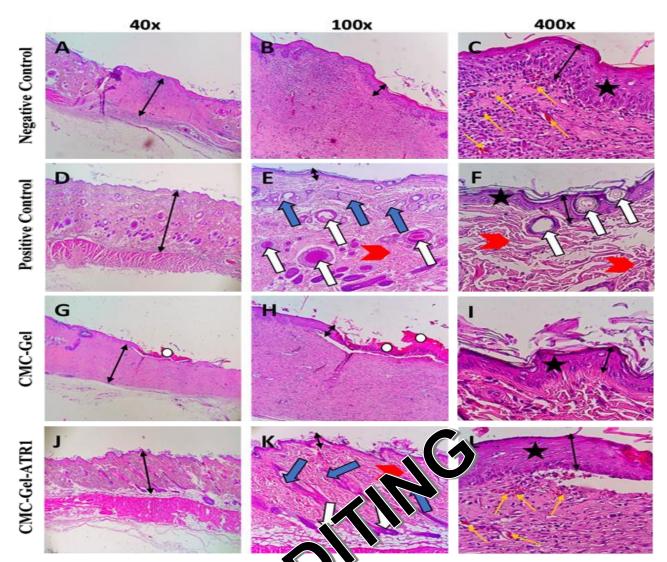


Fig 5. Microscopic sections of the healed incision work of ained with H & E on day 14. Negative control (A, B, C); positive control (D, E, F); CMC-Gel hydrogel (G, H, I); CMC-Gel R1, h) (J, K, L). Black stars: epithelialization, white arrows: hair follicle, blue arrows: sebaceous glands, red arrowheads: many olleger, yellow arrows: neovasculation.

accelerates wound closure in luces inflammation. Histological outcomes in that the CMC-Gel-ATR1 is more similar the surve control in terms of organized epither and scularization, appendage formation, and luage maturity.

ATR shows comble promise for wound healing; however, caution is warranted regarding the disadvantages and limitations noted in the research, particularly the formulation of appropriate drug dosages. This study was limited to full-thickness cutaneous wounds and did not address chronic or complex wound types, such as burns, pressure ulcers, or diabetic wounds, which represent clinical challenges.

CONCLUSION

In this study, CMC-Gel and CMC-Gel-ATR hydrogels were successfully synthesized and evaluated as promising wound dressing compounds. Their physicochemical characteristics, cell viability, and in vivo wound healing properties were examined and supported by histological analysis. Morphological results confirmed the existence of a porous structure consisting of interconnected pores as well as loaded ATR, which are favorable for cell attachment and migration. The CMC-Gel-ATR hydrogel showed desirable properties for wound healing, including moisture absorption, biodegradability, desirable pH, BCI, and blood compatibility. Moreover, both ATR-free

Wound Healing by ATR Mousavi et al.

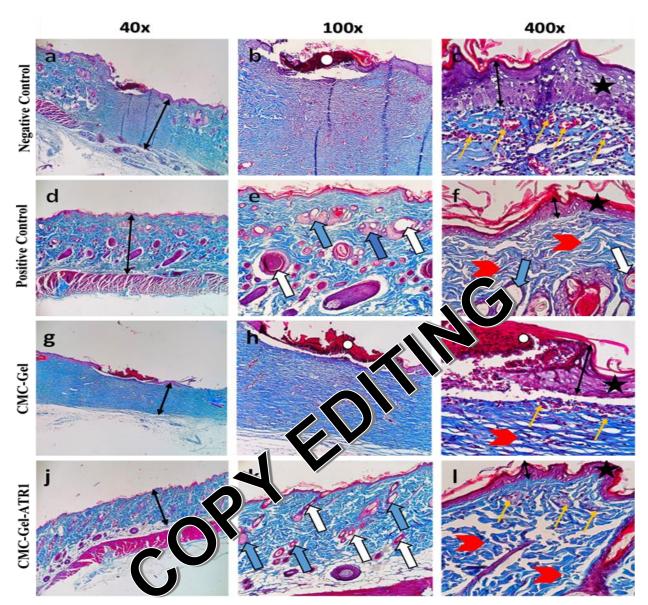


Fig 6. Microscopic sections of with the healed incision wounds stained with Masson's trichrome on day 14. Negative control (A, B, C), positive control (D, E, F); CMC-Gel hydrogel (G, H, I); CMC-Gel-ATR1 hydrogel (J, K, L). Black stars: epithelialization, white arrows: hair follicle, blue arrows: sebaceous glands, red arrowheads: mature collagen, yellow arrows: neovasculation.

and ATR-loaded hydrogels showed no cytotoxic effects on NIH/3T3 fibroblast cells. Animal studies in rat models showed that skin wounds treated with both hydrogels accelerated wound healing within 7 and 14 days. Histological results demonstrated that the hydrogel wound dressings significantly promoted organized epithelialization, vascularization, and collagen maturity, thus accelerating the general wound healing process. Overall, CMC-Gel-ATR wound dressing has many desirable properties for wound healing in clinical applications.

DECLARATIONS

Acknowledgments

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Ethical approval

Animal experiments were approved by the Ethics Committee of Ahvaz Jundishapur University of Medical Sciences, Ahvaz, Iran (ethical code: IR.AJUMS.REC. 1402.513). All experimental procedures were carried out in accordance with the guidelines set forth by the Ethical Standards Committee of Shahroud Medical University, Shahroud, Iran (ethical code: IR.SHMU. REC.1398.157).

Consent to participate

Not applicable.

Consent for publication

All authors have approved the manuscript and agree with its submission to IBJ.

Authors' contributions

SRM: performed experiments, evaluated the data, discussed the results and strategy, and prepared the article draft; MR, AK and MK: contributed in a part of the project management; MS designed and managed the study; SZ: performed experiments, evaluated the data, discussed the results and strategy, and prepared the article draft; GM: designed and managed the study.

Data availability

All relevant data can be found within the map

Competing interests

The authors declare no conflict of inte

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Supplementary information

The online version does not contain supplementary material.

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